

2018 Pr

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#### SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### 1.1 Product identifier

Commercial Product Name Sulfuric acid 94-97% Chemical name: Sulphuric acid

#### Registration number:

01-2119458838-20

1.2 Relevant identified uses of the substance or mixture and uses advised against Use of the Substance/Mixture
 Intermediate, Processing aid, Surface treatment, Catalyst, Acidifier.
 Recommended restrictions on use
 There are no uses advised against.

#### 1.3 Details of the supplier of the safety data sheet

-Company:	Goulding Chemicals Ltd.
-Address:	Centre Park Road, Marina, Cork, Ireland
-Telephone:	+353 (021) 4911611
-Fax:	+353 (021) 4911660
-Contact Email	larry.egar@gouldings.ie

#### 1.4 Emergency number

-Emergency telephone number (outside of office hours): +353 (021) 4911619

## **SECTION 2: Hazards Identification**

#### 2.1 Classification of the substance or mixture

Classification according to Regulation (EU) 1272/2008(CLP) Skin corrosion; Category 1A; Causes severe skin burns and eye damage.

Classification according to EU Directives 67/548/EEC or 1999/45/EC Corrosive; Causes severe burns.

#### 2.2 Label elements

Labelling (REGULATION (EC) No 1272/2008)

Hazard pictograms



Signal word

: Danger



Hazard statements : H314 Causes severe skin burns and eye damage. **Precautionary statements** : Prevention: P280 Wear protective gloves/ protective clothing/ eye protection/ face protection. **Response:** P303 + P361 + IF ON SKIN (or hair): Remove/ Take off P353 immediately all contaminated clothing. Rinse skin with water/ shower. P305 + P351 + IF IN EYES: Rinse cautiously with water for P338 several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P310 Immediately call a POISON CENTER or doctor/physician. Storage: P405 Store locked up.

Hazardous components which must be listed on the label: 7664-93-9 Sulfuric acid

Further information : Never add water to this product.

#### 2.3 Other hazards

Advice; Reacts strongly with water, releasing large amounts of heat.

**Remarks;** This substance is not considered to be persistent, bioaccumulating and toxic (PBT). This substance is not considered to be very persistent and very bioaccumulating (vPvB).

#### **SECTION 3: Composition/Information On Ingredients**

#### 3.1 Substances

Chemical Name Sulphuric acid CAS-No. EINECS-No. / ELINCS No. 7664-93-9 231-639-5

Concentration [%] > 51

Print Date 01.02.2018

#### **SECTION 4: First Aid Measures**

#### 4.1 Description of first aid measures

#### Inhalation

Move to fresh air. Keep patient warm and at rest. Oxygen or artificial respiration if needed. Call a physician if symptoms occur.

#### Skin contact



Print Date 01.02.2018

Wash off immediately with plenty of water removing all contaminated clothes and shoes. Call a physician if symptoms occur.

## Eye contact

Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Obtain medical attention.

#### Ingestion

Rinse mouth. Give small amounts of water to drink. Do NOT induce vomiting. Obtain medical attention.

#### 4.2 Most important symptoms and effects, both acute and delayed

Symptoms : Causes severe burns.

#### 4.3 Indication of any immediate medical attention and special treatment needed

Treatment

: Rinse with plenty of water.

## **SECTION 5: Firefighting Measures**

5.1	Extin	guishing	media
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Extinguishing media

: Sand/Dry powder

Unsuitable extinguishing media

: Do not use a powerful water stream as it may cause corrosive liquid to splash.

# 5.2 Special hazards arising from the substance or

mixture Hot acid splashes.

Heating can release hazardous gases.

#### 5.3 Advice for firefighters

In the event of fire, wear self-contained breathing apparatus. Splashproof protective suit.

#### 5.4 Specific methods

The product itself does not burn. Cool containers/tanks with water spray.

#### **SECTION 6: Accidental Release Measures**

#### 6.1 Personal precautions, protective equipment and emergency procedures

Keep people away from and upwind of spill/leak. Avoid contact with skin and eyes. Wear personal protective equipment. Do not add water into strong acid (risk of splashes).

#### 6.2 Environmental precautions

Should not be released into the environment. Dam up. Soak up with inert absorbent material (e.g. sand, acid binder). Do not use sawdust or inflammable substance.

#### 6.3 Methods and materials for containment and cleaning up



Print Date 01.02.2018

Neutralize with limestone powder and flush with plenty of water. Do not rinse acid into drains that might contain sulphides. Wear personal protective equipment. Dispose of as special waste in compliance with local and national regulations.

# SECTION 7: Handling and Storage

## 7.1 Precautions for safe handling

Handle and open container with care. Do not add water into strong acid (risk of splashes). Wear personal protective equipment. Provide sufficient air exchange and/or exhaust in work rooms. In case of insufficient ventilation, wear suitable respiratory equipment. Ensure that eyewash stations and safety showers are close to the workstation location.

#### 7.2 Conditions for safe storage, including any incompatibilities

Keep in a dry, cool and well-ventilated place. Store in original acid resistant container. Keep away from open flames and hot surfaces. Protect from sunlight.

#### Materials to avoid:

Flammable materials, Bases, chromates, chlorates, nitrates, Sulphides, Oxidizing agents Paper and

cotton carbonize quickly on the influence of sulphuric acid and can catch fire.

#### 7.3 Specific end use(s)

Not applicable

#### **SECTION 8: Exposure Controls/Personal Protection**

#### 8.1 Control parameters

# 8.1.1 Limit values

# Sulfuric acid

OELV - 8 hrs (TWA) = 1 mg/m<sup>3</sup>, : Where no specific short-term exposure limit is listed, a figure three times the long-term exposure limit value should be used

#### 8.2 Exposure controls

#### 8.2.1 Appropriate engineering controls

Avoid contact with skin, eyes and clothing. Wash hands before breaks and immediately after handling the product. Ensure that eyewash stations and safety showers are close to the workstation location.

# 8.2.2 Individual protection measures, such as personal protective equipment Hand protection

Glove material: butyl-rubber, Break through time: 8 h, > 70% sulphuric acid Glove material: Polyethylene, Break through time: 8 h, > 70% sulphuric acid Break through time: 4 h, > 70% sulphuric acid

#### Eye protection

Tightly fitting safety goggles and face-shield.

#### Skin and body protection

# 🚮 GI Chemicals

# Safety Data Sheet according to Regulation (EU) No. 1907/2006 SULPHURIC ACID 94-97%

Revision 4.0

Revision Date 01.02.2018

Print Date 01.02.2018

Protective suit If splashes are likely to occur, wear: apron and boots

# **Respiratory protection**

In case of insufficient ventilation wear suitable respiratory equipment. (filter P3)

#### 8.2.3 Environmental exposure controls

Prevent product from entering the environment.

#### **SECTION 9: Physical And Chemical Properties**

## 9.1 Information on basic physical and chemical properties

General Information (appearance, odour)

Physical state	liquid, oily
Colour	colourless,
	clear
Odour	slightly
	pungent

Important health safety and environmental information

pH Freezing point : Boiling point/boiling range	< 1 ca29 °C 75% H <sub>2</sub> SO <sub>4</sub> ca1 °C 98% H <sub>2</sub> SO <sub>4</sub> ca. 180 °C 75% H <sub>2</sub> SO <sub>4</sub>
Boiling point/boiling range Flash point	ca. 327 °C 98% H₂SO₄ Not applicable
Explosive properties: Lower explosion limit Upper explosion limit	Not applicable Not applicable
Vapour pressure	0,17 hPa ( 20 °C) 75% H₂SO₄ 0,0121 Pa ( 20 °C) 98% H₂SO₄
Density	1,7 g/cm³ ( 20 °C)75% H <sub>2</sub> SO <sub>4</sub> 1,8 g/cm³ ( 20 °C)98% H <sub>2</sub> SO <sub>4</sub>
Solubility(ies):	
Water solubility	completely soluble
Partition coefficient: n-octanol/water	inorganic compound

# GI Chemicals

# Safety Data Sheet according to Regulation (EU) No. 1907/2006 SULPHURIC ACID 94-97% Revision 4.0 Revision Date 01.02.2018

Print Date 01.02.2018

Thermal decomposition	> 300 °C
Viscosity: Viscosity, dynamic	ca. 16 mPa.s (20 °C) 75% H <sub>2</sub> SO <sub>4</sub>
	ca. 30 mPa.s $\ ($ 20 °C) 98% $H_2SO_4$
Oxidising	Not oxidizing
9.2 Other data	
Surface tension	not determined

#### **SECTION 10: Stability And Reactivity**

#### 10.1 Reactivity

Exothermic reaction with water. Do not add water into strong acid (risk of splashes).

#### **10.2 Chemical stability**

Stable under normal conditions.

#### 10.3 Possibility of hazardous reactions

Hazardous reactions	: Reacts with sulphide forming hydrogen sulphide, H <sub>2</sub> S.
	Gives off hydrogen by reaction with metals.
	The forming of hydrogen gas in a closed space
	causes a danger of explosion.

# 10.4 Conditions to avoid

# Conditions to avoid

: High temperatures.

## 10.5 Incompatible materials

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- : Bases chromates
- : nitrates
- : Sulphides
- : Oxidising agents

: Paper and cotton carbonize quickly on the influence of sulphuric acid and can catch fire.

# **10.6 Hazardous decomposition products**

Hazardous decomposition	: sulphuric acid vapour
products	sulphur dioxide



Revision 4.0

Print Date 01.02.2018

Thermal decomposition : >300 °C

#### **SECTION 11: Toxicological Information**

#### 11.1 Information on toxicological

#### effects

#### Acute toxicity

#### Sulfuric acid:

LD50/Oral/Rat: 2 140 mg/kg LC50/Inhalation/4 h/Rat: 0,375 mg/l

Remarks: aerosol

Although the LC50 values from the various inhalation toxicity studies performed with sulphuric acid theoretically trigger classification for Acute inhalation toxicity, classification is not proposed. The effects of sulphuric acid following inhalation are entirely due to local irritation of the respiratory tract: there is no evidence for the systemic toxicity of sulphuric acid in any study, as effects are limited to the site of contact. Classification for acute inhalation toxicity is not considered to be appropriate.

#### Irritation and corrosion

#### Sulfuric acid:

Skin: Corrosive Causes severe burns.

Eves: Corrosive Risk of serious damage to eyes.

#### Sensitisation

Sulfuric acid: Not sensitizing.

#### Long term toxicity

Sulfuric acid: Repeated dose toxicity: Inhalation/Rat/28 d: NOAEL: = 0,0003 mg/l

Carcinogenicity Inhalation/Rat: Did not show carcinogenic effects in animal experiments.

Oral/Mouse: Weak local carcinogen.

Mutagenicity mammalian cells (CHO)/Chromosome aberration test in vitro: Result: positive Metabolic activation: with and without Due to its pH.



Print Date 01.02.2018

Salmonella typhimurium (bacterium)/Ames test: Result: negative Metabolic activation: with and without Reproductive toxicity /Rabbit/Developmental toxicity test: NOEL: = 0,020 mg/l Did not show teratogenic effects in animal experiments.

#### **SECTION 12: Ecological Information**

#### 12.1 Toxicity

#### Aquatic

#### toxicity

Remarks: May be harmful to aquatic organisms because of the low pH value.

#### Sulfuric acid:

LC50/96 h/Lepomis macrochirus (bluegill sunfish)/static test: 16 - 28 mg/l fresh water EC50/48 h/Daphnia magna (Water flea)/static test/OECD Test Guideline 202: > 100 mg/l fresh water EC50/72 h/Desmodesmus subspicatus (green algae)/static test/OECD Test Guideline 201: > 100 mg/l Remarks: May be harmful to aquatic organisms because of the low pH value.

#### Toxicity to other organisms

#### Sulfuric acid:

NOEC/37 d/active sludge/static test: 26 g/l fresh water

NOEC/30 d/active sludge/static test: > 30 g/l fresh water

#### 12.2 Persistence and degradability

# Biological degradability: Sulfuric acid:

The methods for determining biodegradability are not applicable to inorganic substances.

#### 12.3 Bioaccumulative potential

Partition coefficient: n-octanol/water: inorganic compound

**Sulfuric acid:** Does not bioaccumulate.

# 12.4 Mobility in soil

#### Mobility

Vapour pressure: 0,17 hPa (20 °C); 75% H<sub>2</sub>SO<sub>4</sub> 0,0121 Pa (20 °C); 98% H<sub>2</sub>SO<sub>4</sub>, Does not evaporate if spilled to the ground.



## Safety Data Sheet according to Regulation (EU) No. 1907/2006 **SULPHURIC ACID 94-97%** Revision 4.0

Revision Date 01.02.2018

Print Date 01.02.2018

Water solubility: completely soluble Surface tension: not determined

Soil moisture enhances mobility. May be partly neutralized in soil, but significant amounts may leach into the groundwater.

#### 12.5. Results of PBT and vPvB assessment

This substance is not considered to be persistent, bioaccumulating and toxic (PBT). This substance is not considered to be very persistent and very bioaccumulating (vPvB).

#### 12.6. Other adverse effects

None known.

#### **SECTION 13: Disposal Considerations**

#### 13.1 Waste treatment methods

Product Solutions with low pH-value must be neutralized before discharge. Acid must not be rinsed into a drain that might have water containing sulphide. Dispose of as special waste in compliance with local and national regulations. Clean container with water. Refer to manufacturer/supplier for information on recovery/recycling.

## **SECTION 14: Transport Information**

14.1 UN number	1830
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Land transport ADR: Description of the goods: 14.2 UN proper shipping name 14.3 Transport hazard class(es) 14.4 Packing group: Classification code: Risk code	SULPHURIC ACID 8 II C1 80
ADR/RID-Labels:	8
Sea transport IMDG:	
Description of the goods:	
14.2 UN proper shipping name	UN1830, SULPHURIC ACID
14.3 Transport hazard class(es):	8
14.4 Packing group:	II
IMDG-Labels:	8
14.5 Environmental hazards:	Not a Marine Pollutant
Air tropport	

Air transport ICAO/IATA:



# Safety Data Sheet according to Regulation (EU) No. 1907/2006 SULPHURIC ACID 94-97%

Revision 4.0Revision Date 01.02.2018Print Date 01.02.2018

Description of the goods	
14.2 UN proper shipping name	UN1830, Sulphuric acid
14.3 Transport hazard class(es):	8
14.4 Packing group:	II
ICAO-Labels:	8
14.8 Special precautions for user	
None known.	

#### **SECTION 15: Regulatory Information**

#### 15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

Other regulations	:Take note of Directive 96/82/EC on the control of major- accident hazards involving dangerous substances. The product belongs to at least one of the categories 1 through 11 mentioned in Annex 1 of the Directive 1996/82/EC concerning the control of major accident hazards.
	1996/82/EC concerning the control of major accident hazards.

#### Notification status

#### **15.2 Chemical Safety Assessment**

#### **SECTION 16: Other Information**

#### Full text of H-Statements referred to under section 3.

H314 Causes severe skin burns and eye damage.

#### Text of R-phrases mentioned in Section 3

R35 Causes severe burns.

#### **Training advice**

Read the safety data sheet before using the product.

#### **Further information**

The information provided in this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

## Sources of key data used to compile the Safety Data Sheet

Regulations, databases, literature, own tests.

#### Additions, Deletions, Revisions

Relevant changes have been marked with vertical lines.

#### End of Safety Data Sheet